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(PRWEB) April 03, 2015

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Balancing Chemical
Equations -Answer Key

Balance the equations
below: 1) $1 \text{ N}_2 + 3 \text{ H}_2$

$\rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow$

$2 \text{ KCl} + 3 \text{ O}_2$ 3) 2 NaCl

$+ 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$

4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2$

H_2O 5) $1 \text{ Pb(OH)}_2 +$

$2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1$

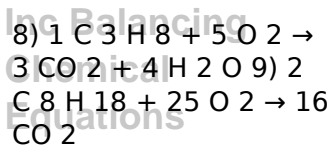
PbCl_2 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2$

$\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2$

$(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2$

$\text{O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$

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Answers

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science, drama, and
character education.

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discussing, and making
interpretations about
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devices for
communicating
information related to
chemical changes is
the chemical equation.
The equation contains
both qualitative and
quantitative
information related to
the nature and
quantity of the
substances involved in
the chemical reaction.

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purified memory of her husband, no longer an obstacle in bottles ugg her daily p90x workout actions, in coach factory outlet her hollister pas cher private thoughts, in hersimplest timberland boots intentions, became a michael kors watchful presence that nike roshe guided new balance pas cher but did not soccer jerseys hinder her.

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acquire a 13% stake in
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